

13E-2ND CLASS

RADIATION

$$q = \epsilon \sigma A (T^4 - T_{\text{sur}}^4)$$

ϵ = emissivity

σ = Stefan-Boltzmann Constant

T_{sur} = Temp. of the surroundings $\neq T_{\text{oo}}$

Black body: $\epsilon = 1$

Grey body: $\epsilon = \alpha$

- All matter emits radiation

- thermal radiation

- in a material the temperature can effect the energy levels of electrons to yield thermal radiation.

- thermal radiation is an **electromagnetic wave**

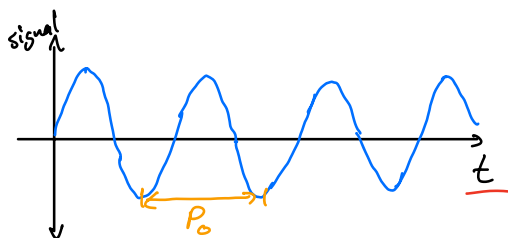
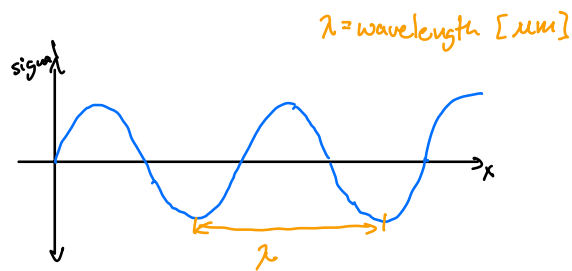
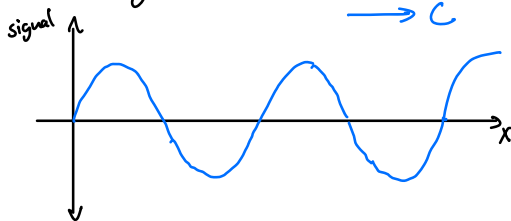
- it travels at the speed of light

$$c = 300 \times 10^6 \text{ m/s}$$

$$= 186,000 \text{ m/s}$$

- it does NOT require a medium

Electromagnetic Waves



$$1 \mu\text{m} = 1 \times 10^{-6} \text{ m}$$

$$1 \text{ nm} = 1 \times 10^{-9} \text{ m}$$

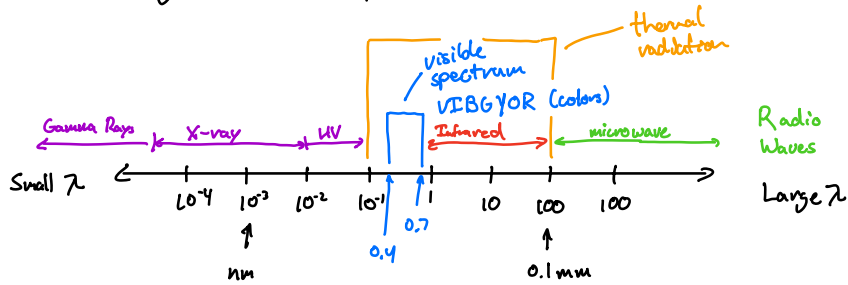
P_0 = period

$$v = \frac{1}{P_0} = \text{frequency} = \frac{1 \text{ cycle}}{\text{second}} = \text{Hz}$$

$c = v \lambda$ ← the speed of the wave is equal to the frequency times wavelength

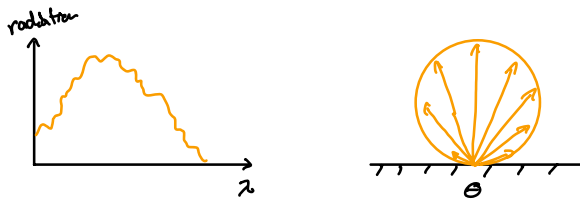
$$\lambda = \frac{c}{v}$$

Electromagnetic [Em] Spectrum



1. λ "spectral"
2. θ "angular, directional"

We will treat radiation as a surface phenomenon



Radiation Fluxes [W/m^2]

E = emissive power = $\epsilon \sigma T_s^4$ - emitted by a surface
 G = irradiation - incident on a surface
 J = radiosity - radiation leaving a surface
 q_{rad} = net radiative flux - net radiation leaving the surface
 $= J - G$

* E, G, J, q_{rad} are over all directions and all the wavelengths